⁵¹V NMR Study on Vanadium(V)-Propylenediaminetetraacetate Complex in Solution

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The solution properties of two isomeric forms of V-PDTA complex(the symmetric α -cis and asymmetric β -cis) have been studied using multinuclear NMR spectroscopy. The $[\alpha$ - $cis]/[\beta$ -cis] ratio was determined by integration of two peaks of 51 V-NMR spectra. The isomerization constant and corresponding thermodynamic parameters were determined under a variable temperature. And the isomerization constant of the aqueous equilibrium mixture was changed by the addition of organic solvents(methanol, formamide, and DMSO).